KHAJA BANDANAWAZ UNIVERSITY

FACULTY OF ENGINEERINGAND TECHNOLOGY, KALABURGI

DEPATMENT OF ENGINEERIG CHEMISTRY

QUESTION BANK FOR BE I AND II SEMESTER

Subject: Engineering Chemistry

Subject Code: 19KBCHE12/22

TWO MARKS QUESTION

MODULE I

- 1. Define electrochemistry with an example
- 2. Define electrode potential & cell
- 3. DefineEMF & give the equation for EMF or E_{cell}
- 4. Define reference electrode with suitable example
- 5. Define concentration cell
- 6. Define battery with an examples

MODULE II

- 1. Define polymerization with an example
- 2. Define degree of polymerization
- 3. Define 1. Homopolymer 2. copolymer with example
- 4. What is number average molecular weight and weight average molecular weight
- 5. Define glass transition temperature (Tg)
- 6. Define conducting polymer
- 7. Define elasticity

MODULE III

- 1. What is corrosion? give an example
- 2. Define electro chemical theory of corrosion
- 3. What is galvanic series
- 4. Define 1. Differential metal corrosion 2. Differential aeration
- 5. What is metal coating and cathodic protection
- 6. Define Sacrificial anode and impressed current method
- 7. Define metal finishing with suitable example
- 8. Define electroplating and electrolessplating with example

MODULE IV

- 1. What is a fuel and give the examples
- 2. Define calorific value
- 3. What are high calorific value and lower calorific value
- 4. What is cracking
- 5. What is reformation of petrol
- 6. Define octane number &Cetane number
- 7. Whatis knocking
- 8. What are anti knocking agent
- 9. Define fuel cell with an example

MODULE V

- 1. What is boiler feed water
- 2. Define scale and sludge formation
- 3. What is priming and foaming
- 4. What is boiler corrosion
- 5. Define dissolved oxygen
- 6. Define biological oxygen demand
- 7. Define chemical oxygen demand
- 8. What is sewage treatment
- 9. What is softening of water
- 10. Define desalination
- 11. What is electro dialysis
- 12. Define nanomaterial with an example
- 13. Define Nano weir and Nano rods
- 14. What are carbon Nano tubes and fullerence

FOUR MARKS QUESTION

MODULE I

- 1. Explain in brief the classification of electrochemical cell with examples
- 2. Discuss in brief the construction and working of calomel electrode
- 3. Explain in brief about the glass electrode and give an equation of Ecell for glass electrode
- 4. Explain the measurement or determination of pH using glass electrode
- 5. Give an EMF equation for concentration cell
- 6. Explain the classification of electrochemical cell
- 7. Write a note batteries
- 8. Give the classification of batteries with an example

MODULE II

- 1. Explain the different types of polymerization
- 2. How molecular weight of polymer is determined
- 3. Discuss in brief the significance of Tg

MODULE III

- 1. Explain differential metal corrosion
- 2. Explain Waterline corrosion & Pitting corrosion
- 3. Explain galvanization process
- 4. Explain tinning process
- Explain in brief the sacrificial anode and impressed current method
- 6. Explain electroplating of nickel (watts bath)
- 7. Explain in brief electro less plating of copper
- 8. Explain manufacture of PCB by electro less plating
- 9. Givethe difference between electro plating and electro less plating

MODULE IV

- 1. Give the classification of fuels
- Explain the synthesis of petrol by Fischer tropsch process
- 3. Explain the reformation of petrol
- Explain the mechanism knocking and harmful effects ofknocking
- 5. Explain about leaded petrol and unleaded petrol
- 6. Explain in brief about power alcohol and biodiesel
- 7. Write the difference between convention cell and fuel cell

MODULE V

- 1. Discuss in brief impurities of water
- 2. Discuss in brief boilercorrosion
- 3. Explain inbrief about priming and foaming and its prevention
- 4. Explain the desalination of sea water by reverse osmosis
- 5. Explain the desalination of water by electro dialysis process
- 6. Explain Nalgonda technique to purify water
- 7. Explain in briefthe classification of nanomaterial's
- 8. Write a note on 1.carbon nano tube 2.fullerence 3. Nano rods 4. Nano wires 5. Dendrimers 6. Nano composites

SIX MARKS QUESTION/EIGHT MARKSQUESTION

MODULE I

- 1. Discuss the construction and working of a glass electrode and application
- 2. Derive Nernst equation for electrode potential
- 3. Explain construction working and application of nickel metal hydride battery
- 4. Describe any three battery characteristics
- 5. Explain primary secondary and reserve batteries with examples
- 6. Describe the construction working of lithium ion batteries mention its application
- 7. How will you determine the electrode potential of unknown electrode using calomel as reference electrode
- 8. Describe the construction and working of zinc air battery. Mention its application
- 9. Explain different types of electrodes
- 10. Explain the determination of pH of given solution using glass electrode
- 11. Numerical problems on concentration cell

MODULE II

- Explain the free radical mechanism of addition polymerization by taking vinyl chloride as an example
- 2. Explain any three factors affecting tg
- 3. Describe the synthesis of 1. Polyurethane 2. Polycarbonate 3.PMMA or plexi glass
- 4. Explain the mechanism of conducting in polyaniline
- 5. Explain any three structure property relationship of polymers
- Numerical problems on number average molecular weight and weight average molecular weight

MODULE III

- 1. Describe the electrochemical theory of corrosion taking rusting of iron as an example
- 2. Mention any six technological importance of metal finishing
- 3. Explain any two factors affecting the rate of corrosion

MODULE IV

- Explain the experimental determination of calorific value of solid/liquid fuel using bomb calorimeter
- 2. Describe the construction and working of solid oxide fuel cell (SOFC) and mention its application
- 3. Explain the construction and working of methanol oxygen fuel cell
- 4. Describe fluidized bed catalytic cracking
- 5. Explain classification of fuel cell based on 1. Temperature 2. Fuel 3. Electrolyte
- 6. Numerical problems on determining calorific value of fuels

MODULE V

- 1. Explain softening of water by ion exchange method
- 2. Explain the activated sludge treatment of sewage water
- 3. Explain thesynthesis of nanomaterial's by sol gel technique
- 4. Illustrate the determination of COD of waste water sample
- 5. Explain size dependent properties of Nanomaterial
- 6. Explain the synthesis of Nano material by chemical vapor deposition method
- 7. Explain wrenkler's method of determining dissolved oxygen
- 8. Explain the prevention of boiler corrosion
- 9. Illustrate the determination of BOD of wastewater sample
- 10. Explain the synthesis of nanomaterial by gas condensation processing
- 11. Explain the synthesis of nanomaterial hydro thermal process
- 12. Numerical problems on COD & BOD